Acta Crystallographica Section C Crystal Structure Communications

ISSN 0108-2701

# Redetermination of ternifoline-C at 150 K

Philip J. Cox et al.

#### **Electronic paper**

This paper is published electronically. It meets the data-validation criteria for publication in Acta Crystallographica Section C. The submission has been checked by a Section C Co-editor though the text in the 'Comments' section is the responsibility of the authors.

© 2000 International Union of Crystallography • Printed in Great Britain - all rights reserved

# electronic papers

Acta Crystallographica Section C Crystal Structure Communications ISSN 0108-2701

# **Redetermination** of ternifoline-C at 150 K

# Philip J. Cox,<sup>a</sup>\* Yanhui Meng,<sup>b</sup> Satyajit D. Sarker,<sup>a</sup> Qinying Deng<sup>c</sup> and Guo Xu<sup>c</sup>

<sup>a</sup>School of Pharmacy, The Robert Gordon University, Schoolhill, Aberdeen AB10 1FR, Scotland, <sup>b</sup>Applied Chemistry, Zhongshan College, Zhongshan 528403, Guandong, People's Republic of China, and <sup>c</sup>School of Chemistry and Chemical Engineering, Zhongshan University, Guangzhong 510275, People's Republic of China

Correspondence e-mail: p.j.cox@rgu.ac.uk

Received 5 September 2000 Accepted 19 September 2000

Data validation number: IUC0000264

The relative stereochemistry of ternifoline-C (7,20-epoxy-*ent*-kaur-16-ene-1,6,7,15-tetrol 1-acetate),  $C_{22}H_{32}O_6$ , previously reported by Wu [*Fudan Xuebao Zir. Kex.* (1988), **27**, 61–65], has been redetermined at 150 K. The molecular geometry and crystal packing agree well with the previous study.

## Comment

The structure of the diterpenoid ternifoline-C, (I), has been reported previously (Wu, 1988) as the *ent*-kaurane but was not included in the Cambridge Structural Database (Allen *et al.*, 1983). We can confirm the structure reported previously, but the cell volume of the present determination is slightly lower [992.4 (3) *versus* 1015 Å<sup>3</sup>] and hence the calculated density higher (1.313 *versus* 1.284 Mg m<sup>-3</sup>), which indicates that the previous structure was probably determined at room temperature.



From the relative stereochemistry shown in the scheme (the absolute stereochemistry has not been determined in this study), the nine chiral centres present in the *ent*-kaurane (assumed) are: C1 S, C5 R, C6 S, C7 S, C8 S, C9 S, C10 S, C13 R and C15 R. If the relative sterochemistry reported is not the absolute sterochemistry, then all these chiral centres will invert to form the kaurane.

The ring conformations (see sketch) approximate to A chair, B boat, C boat, and D half-chair (A is the left-most ring in the scheme). Also, the two six-membered rings formed by the C20 and O1 bridging atoms under the B ring both adopt approximate boat conformations.

Details of three well defined hydrogen bonds (two interand one intramolecular) are shown in Table 2. The C–OH bond lengths are affected by the hydrogen bonds, with C7– O3 being shorter than both C6–O2 and C15–O4, as O3 is only involved in one hydrogen bond, whereas O2 and O4 are each involved in two hydrogen bonds. Endocyclic angle distortions and some torsion angles which help define ring conformations are included in Table 1. Other short nonbonded separations include O2···O3 of 2.772 (4) Å and O4···O3<sup>i</sup> of 2.974 (4) Å [symmetry code: (i) -x,  $y - \frac{1}{2}$ , -z], where the H···O separations are 2.5 Å and the angle subtended by the H atom is less than 120°.

A crystallographic study of a similar fused-ring system has been reported for oridonin (Shi *et al.*, 1992) and similar systems have been studied by NMR (Hou *et al.*, 2000). *ent*-Kauranes (crystalline and amorphous) which lack the C20 and O1 bridging atoms have also been widely reported (Tazaki *et al.*, 1999; Zhao *et al.*, 1999).

## Experimental

The diterpene 7,20-epoxy-*ent*-kaur-16-ene-1,6,7,15-tetrol 1-acetate was extracted from the Chinese medicinal plant *Rabdosia serra* (Maxim.) Hara (Labiatae). A previous study (Wu, 1988) had isolated the diterpene from another species of *Rabdosia*.

Crystal data

$C_{22}H_{32}O_6$	$D_x = 1.313 \text{ Mg m}^{-3}$
$M_r = 392.48$	Mo $K\alpha$ radiation
Monoclinic, $P2_1$	Cell parameters from 3394
a = 10.9870 (16)  Å	reflections
b = 6.0330 (8)  Å	$\theta = 1.91-27.54^{\circ}$
c = 15.429 (3)  Å	$\mu = 0.094 \text{ mm}^{-1}$
$\beta = 103.989 \ (6)^{\circ}$	T = 150 (2)  K
$V = 992.4 (3) \text{ Å}^3$	Needle, colourless
Z = 2	$0.15$ $\times$ 0.05 $\times$ 0.02 mm

Data collection

```
Enraf–Nonius KappaCCD area-
detector diffractometer \varphi and \omega scans to fill Ewald sphere
Absorption correction: multi-scan
(SORTAV; Blessing, 1995)
T_{\rm min} = 0.987, T_{\rm max} = 0.998
7295 measured reflections
```

## Refinement

Refinement on  $F^2$  $w = 1/[\sigma^2]$  $R[F^2 > 2\sigma(F^2)] = 0.059$ where $wR(F^2) = 0.143$  $(\Delta/\sigma)_{max}$ S = 0.967 $\Delta\rho_{max} =$ 3394 reflections $\Delta\rho_{min} =$ 268 parametersAbsoluteH atoms treated by a mixture of<br/>independent and constrained<br/>refinementFlack par

3394 independent reflections 2168 reflections with  $I > 2\sigma(I)$  $R_{int} = 0.074$  $\theta_{max} = 27.54^{\circ}$  $h = -14 \rightarrow 13$ 

 $k = -5 \rightarrow 7$  $l = -18 \rightarrow 18$ 

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0484P)^2] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} &< 0.001 \\ \Delta\rho_{\text{max}} &= 0.214 \text{ e } \text{Å}^{-3} \\ \Delta\rho_{\text{min}} &= -0.255 \text{ e } \text{Å}^{-3} \\ \text{Absolute structure: Flack (1983),} \\ 1054 \text{ Friedel pairs} \\ \text{Flack parameter} &= 1.7 (16) \end{split}$$

Table 1Selected geometric parameters (Å, °).

O1-C7	1.442 (4)	O4-C15	1.428 (4)
O1-C20	1.451 (4)	C6-C7	1.519 (5)
O2-C6	1.438 (4)	C8-C9	1.566 (5)
O3-C7	1.403 (4)	C16-C17	1.319 (6)
C7-O1-C20	112.4 (3)	C9-C10-C5	105.7 (3)
C19-C4-C5	114.7 (3)	C12-C11-C9	113.6 (3)
C4-C5-C10	119.0 (3)	C8-C14-C13	100.6 (3)
C14-C8-C15	99.1 (3)	C15-C16-C13	107.7 (3)
C10-C5-C6-C7	11.3 (4)	C11-C12-C13-C14	21.7 (4)
C14-C8-C9-C11	-8.8(4)	C7-O1-C20-C10	-5.3(4)
C7-C8-C9-C10	-4.7 (4)		

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$\begin{matrix} O2-H2\cdots O4^i\\ O3-H3\cdots O4^{ii}\\ O4-H4\cdots O2 \end{matrix}$	0.76 (6) 0.92 (4) 0.87 (5)	2.07 (5) 1.91 (5) 1.84 (5)	2.795 (4) 2.687 (4) 2.665 (5)	158 (5) 142 (3) 158 (4)
<b>0</b> · · · · · · · · · · · · · · · · · · ·	1			

Symmetry codes: (i)  $-x, \frac{1}{2} + y, -z$ ; (ii) x, 1 + y, z.

The hydroxy H atoms were refined freely and the remaining H atoms were initially placed in calculated positions and thereafter allowed to ride on their attached atoms with common isotropic displacement parameters which converged to 0.034 (2) (non-methyl H atoms) and 0.063 (4)  $Å^2$  (methyl H atoms). The absolute configuration could not be established by this experiment in view of the X-ray wavelength and the elements present.

Data collection: *DENZO* (Otwinowski & Minor, 1997) and *COLLECT* (Hooft, 1998); cell refinement: *DENZO* and *COLLECT*; data reduction: DENZO and *COLLECT*; program(s) used to solve structure: *SIR*97 (Altomare *et al.*,1999); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); software used to prepare material for publication: *PLATON* (Spek, 1998) and *SHELXL*97.

We thank the EPSRC National Crystallographic Service at Southampton University for the X-ray data collection. We wish to acknowledge the use of the EPSRC's Chemical Database Service at Daresbury (Fletcher *et al.*, 1996).

# References

- Allen, F. H., Kennard, O. & Taylor, R. (1983). Acc. Chem. Res. 16, 146-153.
- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115–119.
- Blessing, R. H. (1995). Acta Cryst. A51, 33-38.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Fletcher, D. A., McMeeking, R. F. & Parkin, D. (1996). J. Chem. Inf. Comput. Sci. 36, 746–749.
- Hooft, R. (1998). COLLECT. Nonius BV, Delft, The Netherlands.
- Hou, A.-J., Yang, H., Jiang, B., Zhao, Q.-S., Lin, Z.-W. & Sun, H.-D. (2000). *Fitoterapia*, **71**, 417–419.
- Otwinowski, Z. & Minor, W. (1997). Methods Enzymol. 276, 307-326.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Shi, J.-Q., Chen, Y.-Z., Wu, Q.-J. & Cheng P.-Y. (1992). *Jiegou Huaxue*, **11**, 471–475.
- Spek, A. L. (1998). PLATON. Utrecht University, The Netherlands.
- Tazaki, H., Iwasaki, T., Nakasuga, I., Kobayashi, K., Koshino, H., Tanaka, M. & Nabeta, K. (1999). *Phytochemistry*, 52, 1427–1430.
- Wu, J. (1988). Fudan Xuebao Zir. Kex. 27, 61-65.
- Zhao, Q.-S., Lin, Z.-W., Jiang, B., Wand, J. & Sun, H.-D. (1999). Phytochemistry, 50, 123–126.